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Anti-Plasmodial Activity of Iron Oxide Nanoparticles Derived from *Moringa Oleifera* Stem Bark and Root Extracts

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Abstract

Malaria, caused by Plasmodium parasites, requires a comprehensive treatment strategy, involving the use of phytochemicals, marine organisms, synthetic substances, and nanomaterials. Iron oxide nanoparticles (IONPs) are increasingly attracting attention in medical research because of their biocompatibility, facile degradation, and minimal toxicity. This work presents the results of biosynthesis of IONPs from the extracts of *Moringa oleifera* stem bark and roots, regions that have been relatively unexplored despite their recognized medicinal properties. Characterization of the IONPs was conducted through SEM, XRD, FTIR, and UV-visible spectroscopy. The UV-Vis spectroscopic analysis exhibited distinctive absorption peaks, with IONPsSTEM peaking at 300 nm and IONPsROOT ranging between 300–400 nm. The FTIR spectra illustrated characteristic bands assigned to Fe-O, C=O, and O-H. Examination of SEM images unveiled agglomerates of flattened spherical and spindle-shaped hematite, corroborated by EDX and XRD data. The antiplasmodial efficacy of both variants of IONPs demonstrated moderate effectiveness against Plasmodium parasites, surpassing the crude extracts of the stem and root extracts, albeit falling short of the efficacy of Artemether.

Keywords: Anti-Plasmodial Activity, Iron Nanoparticles, Stem Bark, Root Extracts, *Moringa Oleifera*

Introduction

Malaria has consistently been a global burden. A recent World Health Organization (WHO) report estimated 249 million global population spread across 85 malaria-endemic countries and territories were affected, a notable 5 million cases surge from the preceding year. The global malaria-related mortality which had been steadily declining before 2022 had a slight increase to 608,000 deaths in 2022. This proportion of cases attributable to *Plasmodium vivax* also affects pregnant women whose infection could result in neonates with low birth weight (1).

Malaria, caused by Plasmodium parasites, unfolds through a complex lifecycle involving transmission and fertilization within hosts (2). Notably, *Plasmodium berghei* in rodent hosts mimics the human malaria lifecycle, infiltrating red blood cells (RBC) similarly. Severe *P. berghei* infection in rodents reflects clinical manifestations observed in humans, including vital-organ dysfunction and anaemia (3). Additionally, *P. vivax* and *P. ovale* infections introduce dormant hypnozoites, potentially causing relapses. While uncomplicated falciparum malaria has a low mortality rate, severe cases can lead to organ dysfunction or increased parasitic burden. Severe malaria manifests with diverse clinical symptoms, including coma and jaundice. Septicemia,

often complicating severe malaria, underscores its nuanced nature, particularly in vulnerable populations (4).

Mitigating these Plasmodium parasites requires a multifaceted approach, including the exploration of various compound classes derived from plants (5), marine organisms (6), microbes (7), synthetics (8), and nanomaterials (9). Plant-derived compounds are rich sources of active biomolecules, such as alkaloids, terpenoids, flavonoids, polyphenols, essential oils, and extracts from medicinal plants like *Cissampelos pareira* (10), *Commiphora* spp (11) and *Neorautanenia mitis* (12). These compounds exhibit diverse pharmacological activities, along with their anti-malarial potential. Similarly, marine-derived compounds, including marine natural products (13), algae extracts (14), sponge-derived compounds (15), and coral-derived compounds (16), hold promising antimalarial properties, utilizing their marine biodiversity in drug discovery. Microbial-derived compounds, such as antibiotics (17), antifungal agents (18), and bacterial (19) and fungal metabolites (20) are repurposed by leveraging the vast biochemical diversity of microorganisms in inhibiting malaria. Synthetic compounds, including analogs of natural products (21), small molecules (16), peptides (22), and polymers (23), provide synthetic versatility and



optimization opportunities for antimalarial drug development.

Among these categories, nanomaterials hold particular significance due to their unique physicochemical properties and potential for targeted drug delivery and enhanced therapeutic efficacy (24). Metal nanoparticles (25), metal oxide nanoparticles (26), carbon-based nanomaterials (27), and lipid-based nanoparticles (28) exhibit distinct advantages in terms of stability, biocompatibility, and tunable surface properties, which can be tailored to optimize drug loading, release kinetics, and biodistribution. The application of nanomaterials in malaria treatment poses several advantages over conventional drug formulations, including improved pharmacokinetics, reduced drug resistance, and enhanced cellular uptake, leading to higher intracellular drug concentrations at the site of action (29). Nanomaterial-based drug delivery systems facilitate the co-delivery of multiple therapeutic agents, allowing for synergistic effects and combinatorial therapies to overcome drug resistance and enhance treatment outcomes (30).

Iron oxide nanoparticles (IONPs), as metal nanoparticles have increasing research interests in medicine because of their biocompatibility, biodegradability, and low toxicity (31,32). Their synthesis through chemical methods, with the co-precipitation approach stand out for its efficiency and high yield (33). IONPs serve as indispensable tools in magnetic resonance imaging (MRI) (34), targeted drug delivery systems (35), cancer immunotherapy (36), and as hyperthermia mediators (37), proffering diverse solutions to complex medical challenges. Iron oxide shows antimicrobial potential and generate reactive oxygen species (ROS) which induce oxidative stress, damage or death on parasitic cell (38). However, ensuring the biocompatibility and safety of iron nanoparticles remains paramount, necessitating a thorough understanding of their properties, including size, surface coating molecules, and functional groups (39).

IONPs are synthesized by chemical methods such as co-precipitation (40), thermal decomposition (41), hydrothermal (42), solvothermal (43), microemulsion (44), sol-gel (45), polyol (46), ultrasonic-assisted (47), and electrochemical techniques (48). These methods, though, give good IONPs yields, they are challenged with issues like particle size control, complex setups, and the use of hazardous reagents. The green synthetic approaches however, offer better eco-friendly, low toxic and sustainable solutions. These methods leverage on renewable resources such as plants (49) and microorganisms (50), making them inherently safer and more sustainable. The plant extracts reduces and stabilizes the iron ion using their metabolites to give IONPs during the synthesis (51).

Leaves, seeds, and fruits of plants are rich in biomolecules and widely reported for their role in the synthesis of IONPs. Leaves, in particular, are popular due to their

abundance, affordability, and availability, providing essential metabolites for metal reduction and nanoparticle capping. Extracts from various plant parts such as *Prunus persica* leaves (52), *Laurus nobilis* leaves (53), *Moringa oleifera* leaves (54), and *Moringa oleifera* seeds (55) have been successfully used to synthesize IONPs, acting as both capping and reducing agents. Although there are fewer studies utilizing extracts from stem bark, and roots for IONPs synthesis, their potential remains significant (56) (57). Despite the bioactivities of *Moringa oleifera* stem bark and roots (58), their use in IONPs synthesis has not been explored. This study is aimed at synthesizing IONPs from *Moringa oleifera* stem bark and root extracts and investigation of their anti-plasmodial activities in an *in vivo* model, addressing the knowledge gap.

Materials and Method

Materials

Iron (III) chloride hexahydrate ($\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$), sodium hydroxide (NaOH), ethanol 99.9% of analytical grade (Molychem products) were obtained from a commercial dealer and used without any additional reagents. All the glassware was washed with deionized water and oven dried. The standard antimalarial drug Artemether was obtained commercially. *Moringa oleifera* stem barks and root were freshly collected from the Medicinal Garden at Sheda Science and Technology Complex (SHESTCO), Abuja. The plants were identified by a botanist, Mr Akeem Lateef and the specimen voucher number (NIPRD/H/7405) was deposited at National Institute of Pharmaceutical Research and Development, NIPRD, Idu, Nigeria. The deionized water was used for all the homogenization process.

Preparation of *Moringa oleifera* Stem Bark and Root Extract

To prepare the extract of *Moringa oleifera* stem barks, the barks were washed with deionized water to remove impurities, air dried and pulverized. The fine powdered *Moringa oleifera* stem bark of 6g was added into 200 mL deionized water which was heated at 100°C for 20 to 30 minutes. The obtained extract was filtered using filter paper (Whatman no. 1) and stored at 4°C for further use (59).

Similar method was used to obtain *Moringa oleifera* root extract.

Synthesis of IONPs of *Moringa Oleifera* Stem Bark and Root Extract

The 50 mL extract of *Moringa oleifera* stem barks was added dropwise with 50 mL of 0.001M (3.244g) $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$ solution and continuously stirred to add 1 M NaOH solution. Brown precipitates appeared which was stirred for 1 hour 30 minutes at 70°C on a magnetic hotplate. The colour change confirmed the formation of IONPs which was centrifuged at 3000 rpm in 20 minutes, washed with ethanol and oven dried at 90°C for 3 hours. The reddish brown powder obtained was calcined at temperature at 450°C for 2 hours (60).



Similar procedure involving the extract of *Moringa oleifera* root was used to obtain other IONPs.

Characterization of Iron Oxide Nanoparticles of *M. oleifera* Stem and Root Extract

The UV-VIS characterization involved JENWAY 6405 UV-Vis spectrophotometer to confirm and characterize the synthesis of IONPs. The UV-Vis absorption spectrum was validated through a wavelength scan spanning from 200 to 800 nm. Having synthesized IONPs through the biogenic method, the UV-Vis spectra of the solution was taken at different wavelengths.

FTIR analysis provides spectroscopic information about bond details of compounds. For these green synthesized IONPs, FTIR spectra were recorded on an Agilent 630 Cary FTIR Spectrometer. This model of spectrometer has two modules - an ATR module for liquids and solids, and a 'Dial Path' module for absorbance spectra of liquids, films and gels. For this analysis, the IONPs were prepared on KBr discs using a 4% (w/w) solid/KBr mixture. The FTIR spectra were acquired in the range of 400–4,000 cm^{-1} . The vibrational mode along with the bond details of compounds was studied. Likewise, the biomolecules present in plant extract at 1 mm thick re-crystallized KBr disc.

The X-ray Diffractometer (XRD) Thermo scientific model: ARL 'XTRA X-ray and serial number 197492086 analyzed the materials. XRD characterization of thin film of each sample was done in Umaru Musa Yar'adua University, Katsina State, Nigeria. As a non-destructive analytical technique for characterizing crystalline materials, information regarding the structures, phases, texture, crystal lattice parameters, chemical composition and other structural parameters such as average grain size, crystallinity, crystal defects and strain could be obtained.

The materials also were analyzed by EDX Scanning Electron Microscope (SEM) model: PRO:X: 800-07334 Phenom World and serial number MVE01570775. This analytical technique gives information regarding the structures, surface morphology of the materials. It was as well performed pore and fibre metric analysis together with particles sizes on the IONPs.

$$\% \text{ Parasite Growth} = \frac{\text{Number of Parasitized Cells} \times 100}{\text{Number of Red Blood Cells (RBC)}} \quad (1)$$

$$\% \text{ Inhibition} = \frac{\% \text{ Parasitaemia}_{\text{control}} - \% \text{ Parasitaemia}_{\text{test group}} \times 100}{\% \text{ Parasitaemia}_{\text{control}}} \quad (2)$$

Statistical Analysis

The collected data from the *in vivo* exercise were analyzed using OriginPro Software® 2017 with the results of study expressed as a mean \pm standard error of the mean ($M \pm SE$). One-way analysis of variance (ANOVA) was used to compare between groups followed by Tukey and Levene post-hoc. Differences of $p < 0.05$ were considered significant.

Experimental Animals

For experimental purpose, Wistar rats obtained from the College of Veterinary Medicine, Joseph Sarwuan Tarka University, Makurdi were housed in a 15x30cm cage with metal tops. The cage was regularly maintained with constant change of the wood shavings (saw dust) serving as bedding materials. The experimental rats grouped into six were acclimatized for 2 weeks. Each of these groups made up of 2 rats was treated separately with the stem bark and root extracts of *Moringa Oleifera* along with their corresponding IONPs as the test groups while the remaining groups had positive and negative controls. The positive control was treated with Artemether tablet. The negative control, however, was neither treated nor induced.

Suppression of Plasmodium Parasites by IONPs

The IONPs suppression of the Plasmodium parasites was deduced along with the extracts of *Moringa Oleifera* stem bark and root (the capping agents), then the positive and negative controls in agreement with some reports (61). About 2mL of fresh blood was collected from the two weeks acclimatized rats (115 – 120 g) and was mixed with 0.6mL of normal saline. The 0.5 mL parasitaemia (1×10^7 *Plasmodium berghei*) in blood saline suspension was administered intraperitoneally as infection on the rats and were left for 24 hours to allow the Plasmodium parasites spread.

Afterwards, treatments by each extract of the stem and roots of *Moringa oleifera* along with their corresponding IONPs and positive control (Artemether) were carried out at 50 mg/kg body weight. After 48 hours, the blood samples from pricked tail of the experimental rats were taken unto a clean glass slide, smeared to make a thin blood film, air-dried and mixed with methanol for 1 hour. Then staining with giesma stain for 10 minutes and rinsed. When dried, each specimen was viewed under the microscope of x100 objective lens (62) (63). The percentage parasite growth and percentage inhibition on the red blood cells (RBC) were determined using the given equations (1) and (2):

Results and Discussion

Biosynthesis of IONPs

The biosynthesis of iron oxide nanoparticles (IONPs) was achieved by reducing and stabilizing iron salts in an aqueous solution using *Moringa oleifera* extracts. The synthesis involved reacting ferric chloride with extracts from the stem and root of *Moringa oleifera*, which resulted in a colour change of the reaction mixture to reddish-brown, indicating the formation of IONPs.



UV-Vis Spectra of Biosynthesized IONPs

Investigating the optical properties of the green synthesized iron oxide nanoparticles (IONPs), UV-Vis absorption spectra were recorded, as shown in Figure 1.

The UV-Vis spectroscopy displayed an absorption peak at 300 nm in the UV region, indicating the formation of IONPs using *Moringa oleifera* stem bark extracts (IONPsSTEM) (64). This peak, due to the quantum

confinement effect, sets these nanoparticles apart from bulk iron oxide materials (65). Importantly, the absence of additional peaks associated with impurities and defects indicates the high purity and quality of the crystalline biosynthesized IONPs from stem extracts. Similarly on the spectra (Figure 1), the UV-Vis absorption confirmed the formation of IONPs from root extracts (IONPsROOT), showing an absorption peak in the range 300 – 400 nm with no extra peaks, which further validates the IONPs were qualitatively pure (66).

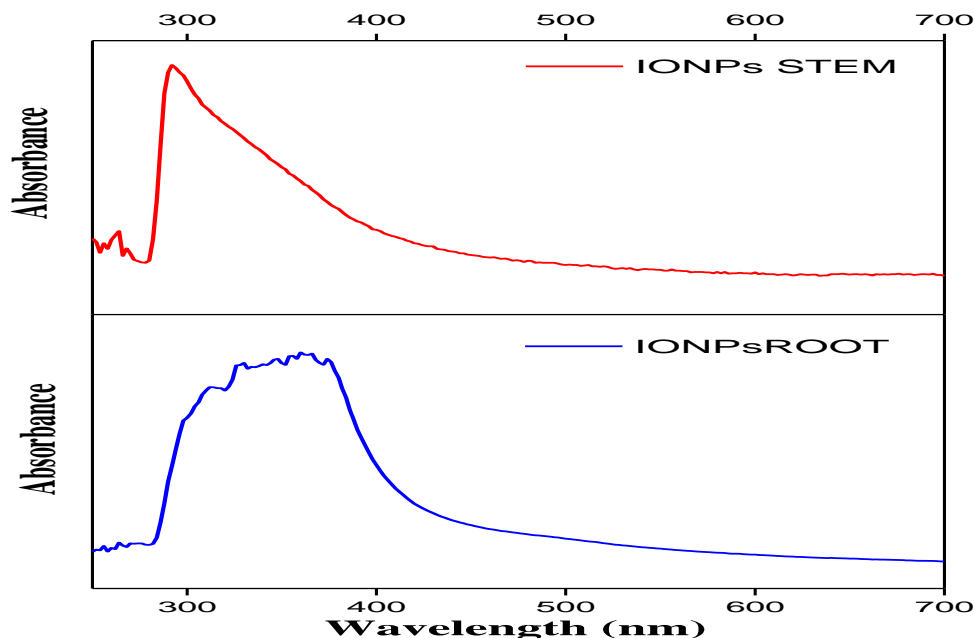


Figure 1: UV-Visible Spectra of IONPsSTEM and IONPsROOT

FTIR Analysis of IONPs

The Fourier Transform Infrared (FTIR) spectroscopic analysis of biosynthesized iron oxide nanoparticles (IONPs) from *Moringa oleifera* stem and root extracts gave absorption bands indicative of specific functional groups. For IONPsSTEM, as shown in Figure 2, the absorption bands at 797 cm^{-1} , 1036 cm^{-1} , and 1394 cm^{-1} were attributed to the stretching vibrations of Fe–O bonds, C–O stretching, and C–H bond stretching, respectively (67). The band at 1610 cm^{-1} was associated with the C=O stretching of amide groups, while the bands at 1893 cm^{-1} , 1990 cm^{-1} , and 2109 cm^{-1} were linked to various functional groups present in the stem extract. Additionally,

the band at 2892 cm^{-1} indicated C–H bond stretching, and the broad band at 3220 cm^{-1} was due to hydroxyl (O–H) stretching. Similarly, the root-derived IONPs in Figure 2b, showed an absorption band at 685 cm^{-1} for Fe–O bond stretching, 1625 cm^{-1} for C=O stretching of amide groups, and bands at 1871 cm^{-1} , 2117 cm^{-1} , and 2340 cm^{-1} for various organic functional groups (68). The bands at 2802 cm^{-1} and 2884 cm^{-1} were associated with C–H bond stretching, and the band at 3250 cm^{-1} was due to O–H stretching. The deduction from the findings suggest the reducing and capping ability of the biomolecules in the *Moringa oleifera* extracts on the iron salts (69).

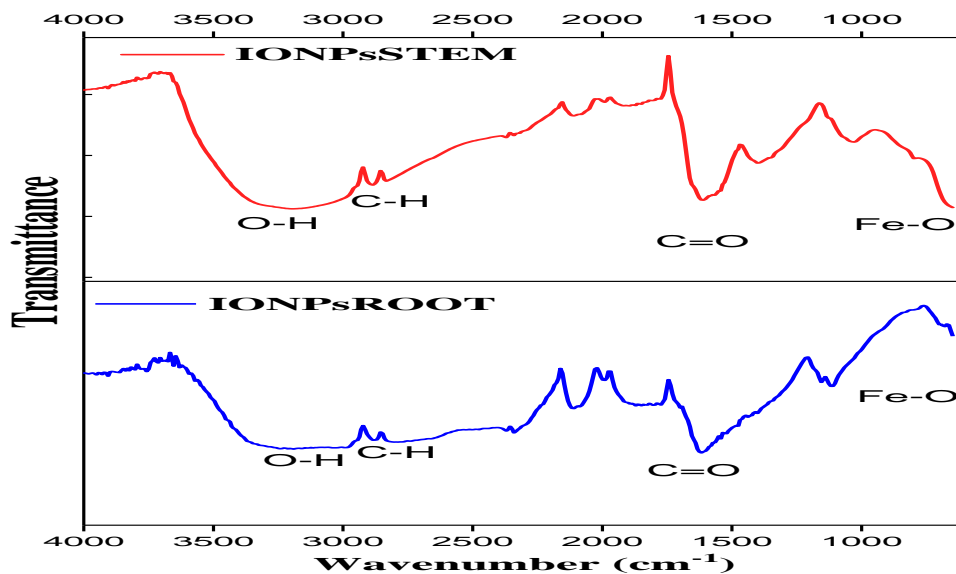
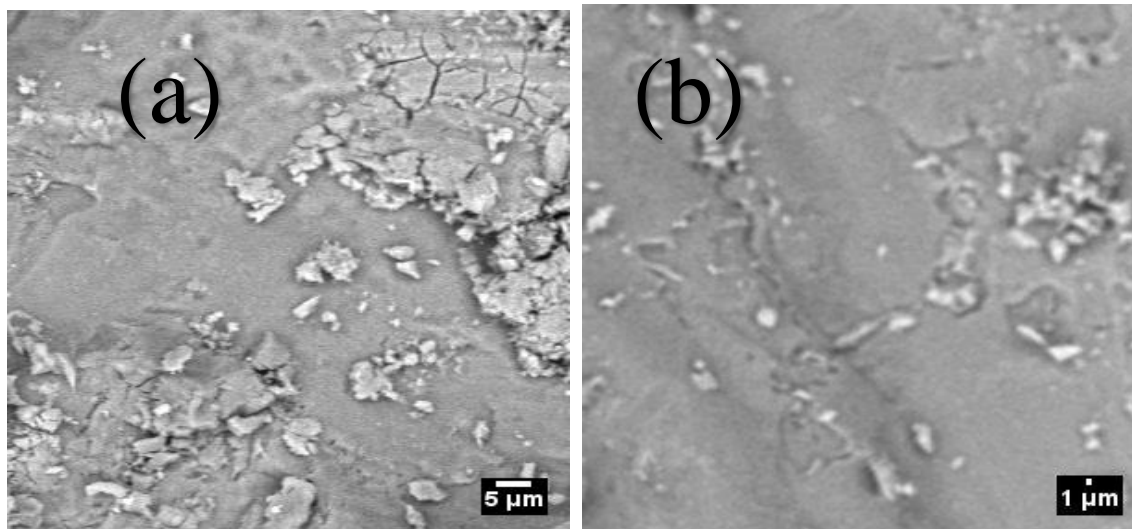


Figure 2: FTIR Spectra of IONPsSTEM and IONPsROOT

SEM Analytical Technique

The surface morphology and size of the synthesized IONPs derived from *Moringa Oleifera* stem bark and root extract were analyzed by SEM. The SEM images in Figures 3a, b, c and d for respective 5 μ m and 1 μ m magnifications. The images revealed an interaction between iron and the

biomolecules of the plant extracts resulting in spherically flattened hematite (70) and spindle hematite shapes (71) respectively. The SEM images are agglomerated with the chunk of particles in micro range, i.e. the presence of bioactive substances originating from plants capping the metal element.(72).



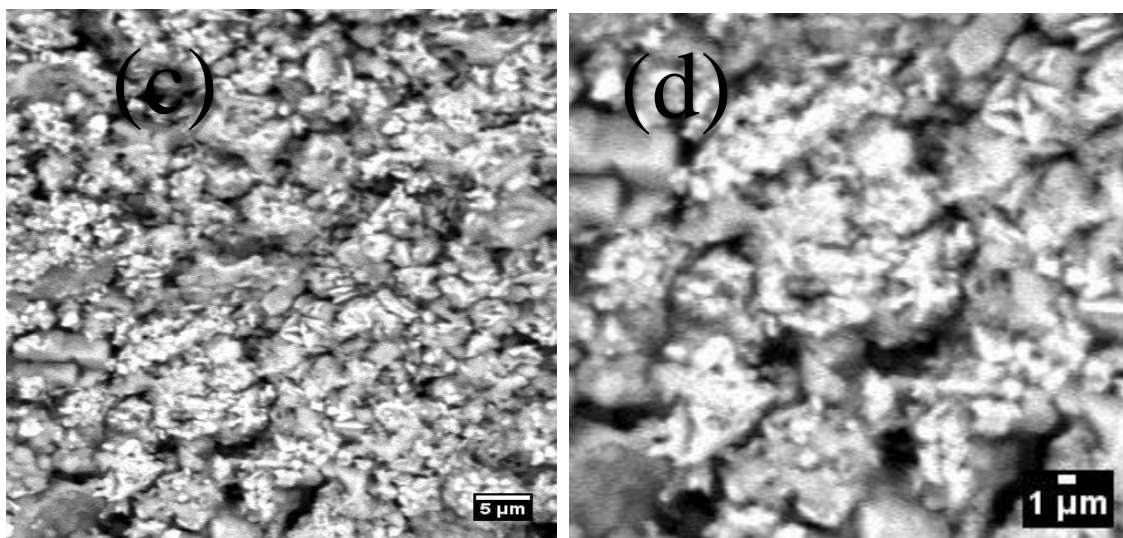


Figure 3: SEM Images of (a) IONPsSTEM at 5µm (b). IONPsSTEM at 1 µm (c) IONPsROOT at 5µm (d) IONPsROOT at 1 µm

EDX Analysis of IONPs

The elemental composition analysis of the iron oxide nanoparticles (IONPs) confirmed the presence of iron as shown in figures 4a and b, affirming the purity of the biosynthesized IONPs. In Tables 1 and 2, the relative percentage of iron in the nanoparticles derived from stem

bark and root were found to be 14.69% and 61.82%, respectively. These values, clearly highlighted in Figures 4a and b, closely align with the theoretically expected stoichiometric mass percentages, indicating the successful synthesis and high purity of the IONPs (73).

Table 1: EDX Weight Ratio of Electrospun IONPsSTEM

Element Number	Element Symbol	Element Name	Atomic Conc.	Weight Conc.
26	Fe	Iron	8.10	14.69
17	Cl	Chlorine	39.24	45.14
11	Na	Sodium	48.92	36.50

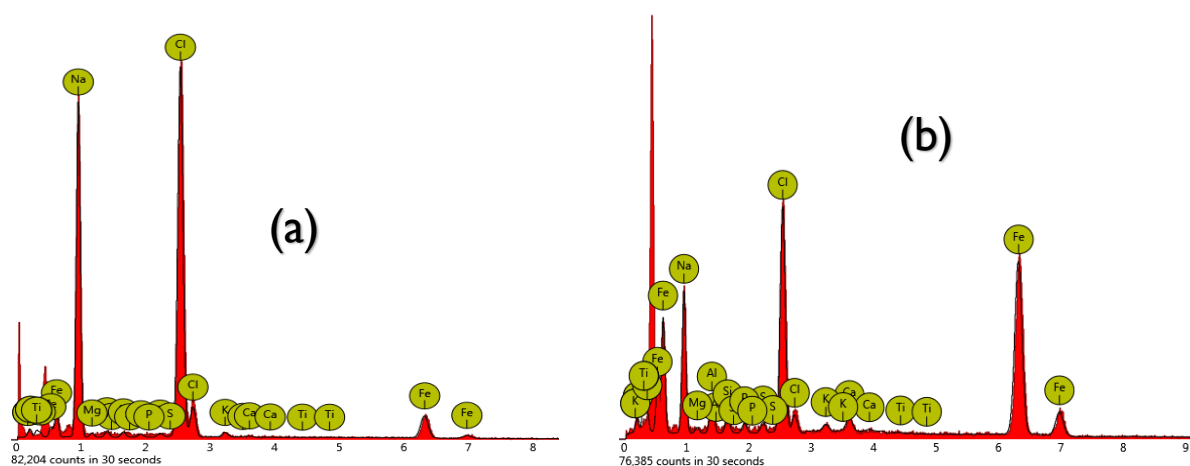


Figure 4: EDX of (a) IONPsSTEM (b) IONPsROOT

**Table 2: EDX Weight Ratio of Electrospun IONPsROOT**

Element Number	Element Symbol	Element Name	Atomic Conc.	Weight Conc.
26	Fe	Iron	45.35	61.82
17	Cl	Chlorine	18.44	15.96
11	Na	Sodium	25.98	14.58

XRD Pattern of IONPs

The Powder XRD patterns of biosynthesized iron oxide nanoparticles (IONPs) derived from both the stem bark and root extracts of *Moringa oleifera* are presented in Figure 5, respectively. The diffraction peaks for the stem bark-derived IONPs can be indexed to the flattened form of hematite (Fe_2O_3) crystals, according to JCPDS card No. 00-002-0837. The observed peak positions at 2θ values of 29.14° , 32.28° , and 45.98° correspond to the crystal planes

(104), (113), and (024), respectively (74). Similarly, the root-derived IONPs showed diffraction peaks consistent with the spindle form of hematite, with peak positions at 2θ values of 24.28° , 33.38° , 35.17° , 38.3° , 42.7° , 45.60° , 49.70° , 54.28° , 57.62° , 58.96° , 64.22° , and 69.94° , corresponding to the crystal planes (012), (104), (110), (006), (113), (202), (024), (116), (112), (214), (300), and (208) respectively (75).

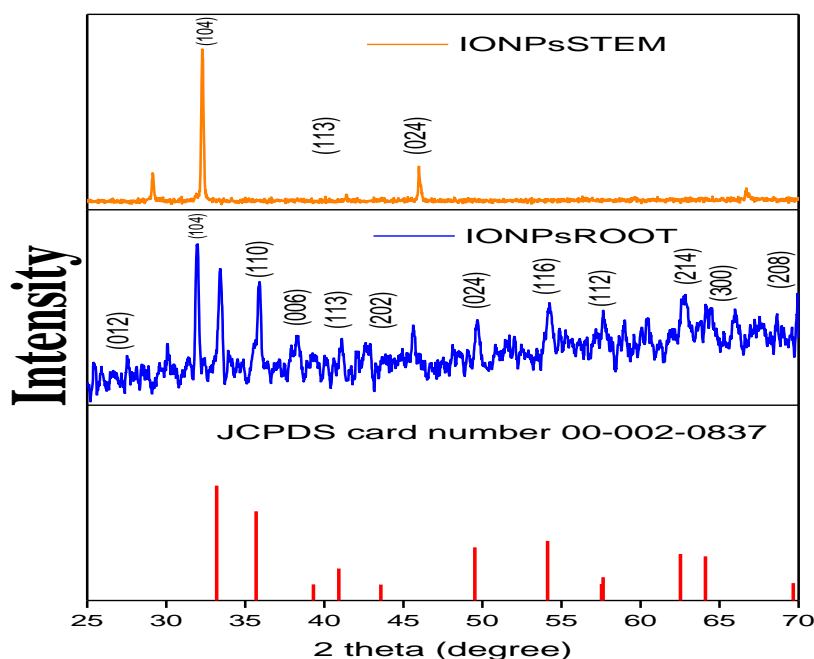


Figure 5: XRD patterns for synthesized IONPsSTEM, IONPsROOT showing different peak positions at 2θ along with the JCPDS card number 00-002-0837 indicating the miller indices of the synthesized nanoparticles.

The crystallite sizes of these IONPs were calculated using the Scherrer equation (3),

$$D = \frac{K\lambda}{\beta \cos\theta} \quad (3)$$

where D is the crystallite size, K is the Scherrer constant (0.9), λ is the X-ray wavelength (0.1541 nm), β is the full width at half maximum (FWHM) of the diffraction peak, and θ is the peak position (76). From Table 3, the stem bark-derived IONPs, the crystallite sizes ranged from

47.86 nm to 62.23 nm, with an average size of 52.91 nm, while root-derived IONPs as shown in Table 4, ranged from 4.43 nm to 38.57 nm, with an average size of 12.96 nm. The peak broadening in the XRD patterns indicates the presence of small nanocrystals. The calculated crystallinities according to equation (4) were 13.98 and 38.86 for the stem bark and root extracts, respectively, further supporting the formation of nanocrystals (77).



$$\text{Crystallinity} = \frac{\text{Area of crystalline peaks}}{\text{Area of all peaks (crystalline+amorphous)}} \times 100 \quad (4)$$

Table 3: Some Crystallographic Parameters of the Synthesized IONPsSTEM based on the X-Ray Diffraction Pattern

S/N	Peak Position (2θ)	FWHM (degree)	(β)	Crystallite size D (nm)	D (Average) (nm)
1	29.14	0.1533		53.55	52.91
2	32.28	0.1728		47.86	
3	45.98	0.1387		62.23	
4	66.70	0.1982		47.99	

Table 4: Some Crystallographic Parameters of the Synthesized IONPsROOT based on the X-Ray Diffraction Pattern

S/N	Peak Position (2θ)	FWHM (β) (degree)	Crystallite size D (nm)	D (Average) (nm)
1	24.28	0.6493	12.5154	12.9666
2	31.96	0.3291	25.1082	
3	33.38	0.4121	20.1274	
4	35.86	0.4919	16.9732	
5	38.30	0.7737	10.8697	
6	41.04	0.7749	10.9469	
7	42.70	1.0659	8.0026	
8	45.60	1.3843	6.2251	
9	49.70	0.5422	16.1461	
10	54.28	0.7660	11.6543	
11	57.62	1.5336	5.9118	
12	58.96	1.1475	7.9528	
13	60.48	2.0740	4.4336	
14	62.78	1.9800	4.7002	
15	64.22	1.2800	7.3274	
16	69.94	0.2514	38.5703	

Antiplasmodial Activity of Biosynthesized IONPs of *Moringa Oleifera* Stem and Root Extract

The anti-plasmodial activity of iron oxide nanoparticles (IONPs) derived from *Moringa oleifera* stem and root extracts as shown in Figures 6 and 7 was evaluated against Plasmodium parasites, showing moderate activity compared to the positive control Artemether (78). *In vivo* tests on experimental rats revealed in Figure 7 that IONPs from the stem extract inhibited parasitaemia by

48.58±1.77%, while the root-derived IONPs exhibited a parasitaemia inhibition of 52.24±6.40% (79). These results as shown in Table 5 were compared to the *M. oleifera* stem and root extracts alone, which showed lower inhibition rates. The study demonstrated the potential of IONPs synthesized from *M. oleifera* extracts as effective anti-malarial agents, with the positive control Artemether exhibiting the highest inhibition rate at 70.45±0.00% (80).

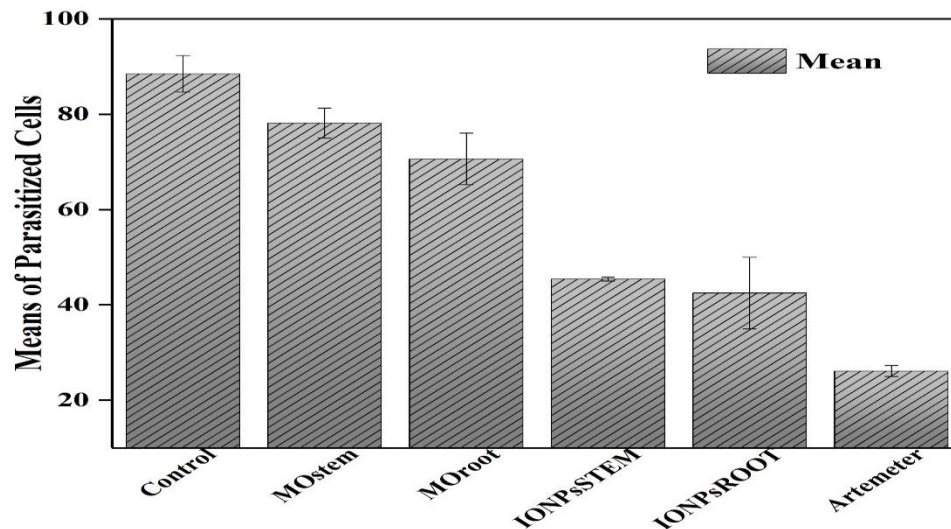


Figure 6: Number of parasitized cells of the Negative Control and Test Groups of Wistar Rats. All Values are presented as mean \pm SEM $p < 0.05$ was considered significant for comparison.

Table 5: Antiplasmodial Activity of IONPs and their Respective *Moringa Oleifera* Stem and Root Extract along with the Positive Control (Artemether) against *Plasmodium berghei*

Samples	Negative Control	<i>M. Oleifera</i> Stem Extract	<i>M. Oleifera</i> Root Extract	IONPsSTEM	IONPsROOT	Artemether
% Inhibition on the Parasites	0	11.36 \pm 7.39	20.30 \pm 2.36	48.58 \pm 1.77	52.24 \pm 6.40	70.45 \pm 0.00

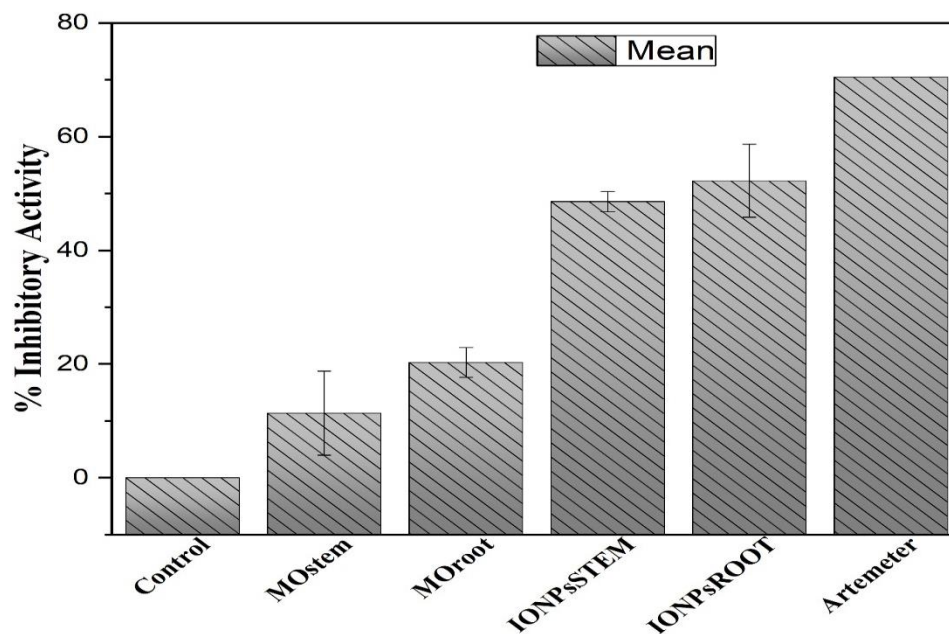


Figure 7: *In vivo* Anti-plasmodial Activity in percentage of IONPsSTEM, IONPsROOT, their *Moringa Oleifera* Stem and Root Extracts, Artemether (Positive Control) and Negative Controlled Rats. Values are presented as mean \pm SEM $p < 0.05$ was considered significant for comparison.



Iron oxide nanoparticles (IONPs) synthesized from the stem and root of *Moringa oleifera* demonstrate strong anti-plasmodial effects, particularly when paired with phytochemicals that boost immune response. Upon entry into infected red blood cells, these IONPs discharge iron ions, prompting oxidative stress crucial for impairing the parasite's nucleus, effectively suppressing Plasmodium proliferation and surpassing the efficacy of these plant extracts alone (79). The significance of iron in combatting pathogens is necessitated by the performance of IONPs derived from diverse plant sources such as *Nephrolepis exaltata*, *Camellia sinensis*, *Matricaria chamomilla* L., *Artemisia herba-alba* Asso, and *Punica granatum* L. peel, which are of notable antimicrobial properties against a range of pathogens, including both gram-positive and gram-negative bacteria, underscoring the importance of iron in inhibiting pathogenic activity (81).

Conclusion

The eco-friendly synthesis of IONPs derived from *Moringa oleifera* stem and root extracts was successfully carried out and

characterized by UV-VIS, FTIR, SEM, EDX, and XRD. These nanoparticles, with average grain sizes of about 52.9 nm and 12.9 nm respectively, demonstrated significant antiplasmodial activity against the Plasmodium parasites, and over the extracts of *Moringa oleifera* stem bark and roots. These results express the potential of the green-synthesized IONPs as effective antimalarial agents. Therefore, it is recommended to synthesize iron nanoparticles from isolated active antiplasmodial compounds from *Moringa oleifera* and other medicinal plants to enhance their therapeutic efficacy. However, further work on toxicity and clinical trials are required on the synthesized nanomaterials. In addition, conducting comprehensive histopathological analyses to assess the toxicity of IONPs in major organs is crucial for ensuring their safety for clinical use. Similar work on other pathogens could further show the versatility and potential of IONPs in treating various infectious diseases

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